

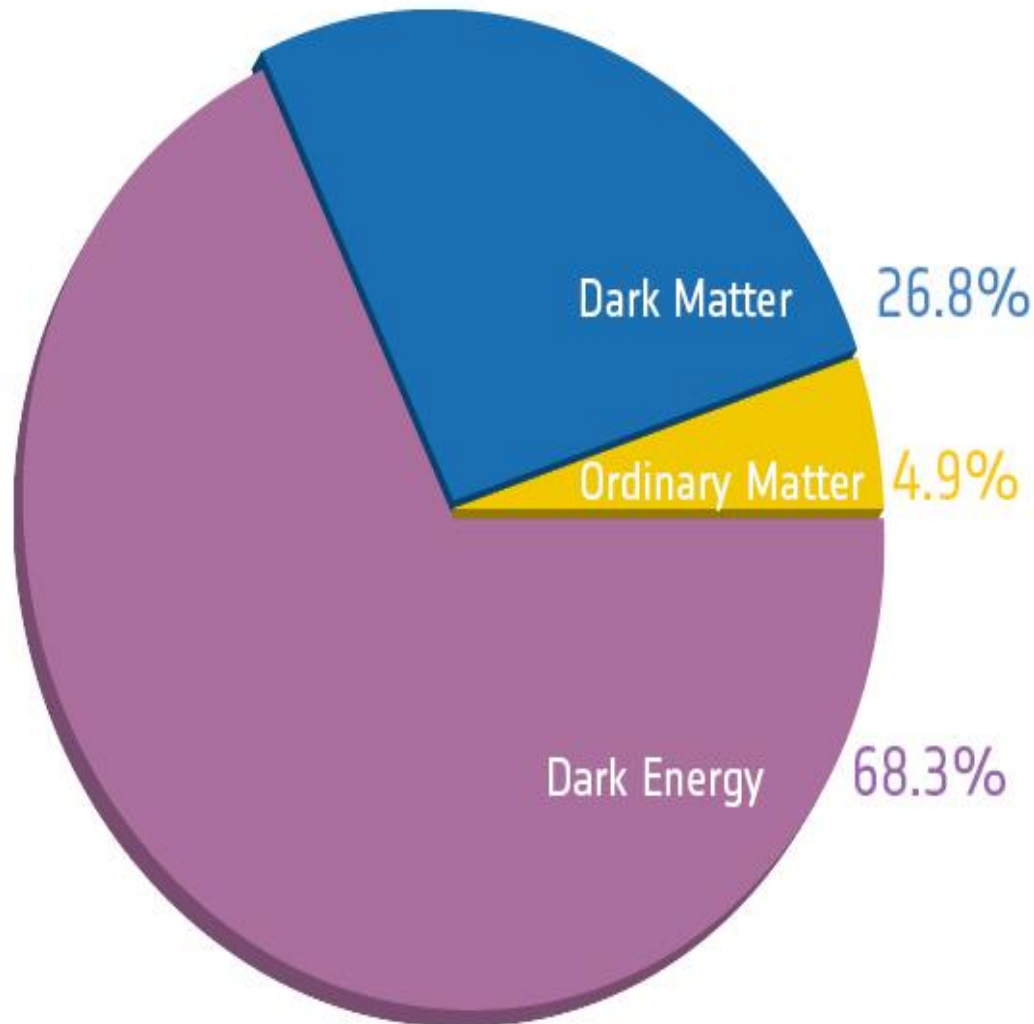
Nuclear and Particle Physics

Εισαγωγή

Τάσος Λιόλιος
Μάθημα Πυρηνικής Φυσικής

Η σύσταση του σύμπαντος

(σύμφωνα με τα δεδομένα του Planck mission-2013)



A standard periodic table of elements, color-coded by groups. It includes element symbols, atomic numbers, and names. The table is organized into blocks: s-block, p-block, d-block, and f-block. It also includes labels for 'Non-Metals', 'Metals', and 'Transition Metals'.

Τα δομικά στοιχεία της συνήθους ύλης
(σε ατομικό επίπεδο - χημικά στοιχεία)

1 New Designation
IA Original Designation

1 **H**
1.0094

2 **He**
4.00260

Non-Metals

Atomic #
Symbol
Atomic Mass

13 **Al**
26.982

14 **Si**
28.086

15 **P**
30.974

16 **S**
32.06

17 **Cl**
35.453

18 **Ar**
39.948

19 **K**
39.098

20 **Ca**
40.08

21 **Sc**
44.956

22 **Ti**
47.88

23 **V**
50.942

24 **Cr**
51.996

25 **Mn**
54.938

26 **Fe**
55.847

27 **Co**
58.933

28 **Ni**
58.69

29 **Cu**
63.546

30 **Zn**
65.39

31 **Ga**
69.72

32 **Ge**
72.59

33 **As**
74.922

34 **Se**
78.96

35 **Br**
79.904

36 **Kr**
83.80

37 **Rb**
85.468

38 **Sr**
87.62

39 **Y**
88.906

40 **Zr**
91.224

41 **Nb**
92.906

42 **Mo**
95.94

43 **Tc**
(98)

44 **Ru**
101.07

45 **Rh**
102.91

46 **Pd**
106.42

47 **Ag**
107.87

48 **Cd**
112.41

49 **In**
114.82

50 **Sn**
118.71

51 **Sb**
121.75

52 **Te**
127.60

53 **I**
126.91

54 **Xe**
131.29

55 **Cs**
132.91

56 **Ba**
137.33

57 **La**
138.91

58 **Ce**
140.12

59 **Pr**
140.91

60 **Nd**
144.24

61 **Pm**
(145)

62 **Sm**
150.36

63 **Eu**
151.96

64 **Gd**
157.25

65 **Tb**
158.93

66 **Dy**
162.50

67 **Ho**
164.93

68 **Er**
167.26

69 **Tm**
168.93

70 **Yb**
173.04

71 **Lu**
174.97

72 **Hf**
178.49

73 **Ta**
180.95

74 **W**
183.85

75 **Re**
186.21

76 **Os**
190.2

77 **Ir**
192.22

78 **Pt**
195.08

79 **Au**
196.97

80 **Hg**
200.59

81 **Tl**
204.38

82 **Pb**
207.2

83 **Bi**
208.98

84 **Po**
(209)

85 **At**
(210)

86 **Rn**
(222)

87 **Fr**
(223)

88 **Ra**
226.03

89 **Ac**
227.03

90 **Th**
232.04

91 **Pa**
231.04

92 **U**
238.03

93 **Np**
237.05

94 **Pu**
(244)

95 **Am**
(243)

96 **Cm**
(247)

97 **Bk**
(247)

98 **Cf**
(251)

99 **Es**
(252)

100 **Fm**
(257)

101 **Md**
(258)

102 **No**
(259)

103 **Lr**
(260)

104 **Unq**
(261)

105 **Unp**
(262)

106 **Unh**
(263)

107 **Uns**
(262)

108 **Uno**
(265)

109 **Une**
(266)

110 **Uun**
(267)

Transition Metals

Metals

Rare Earth Elements

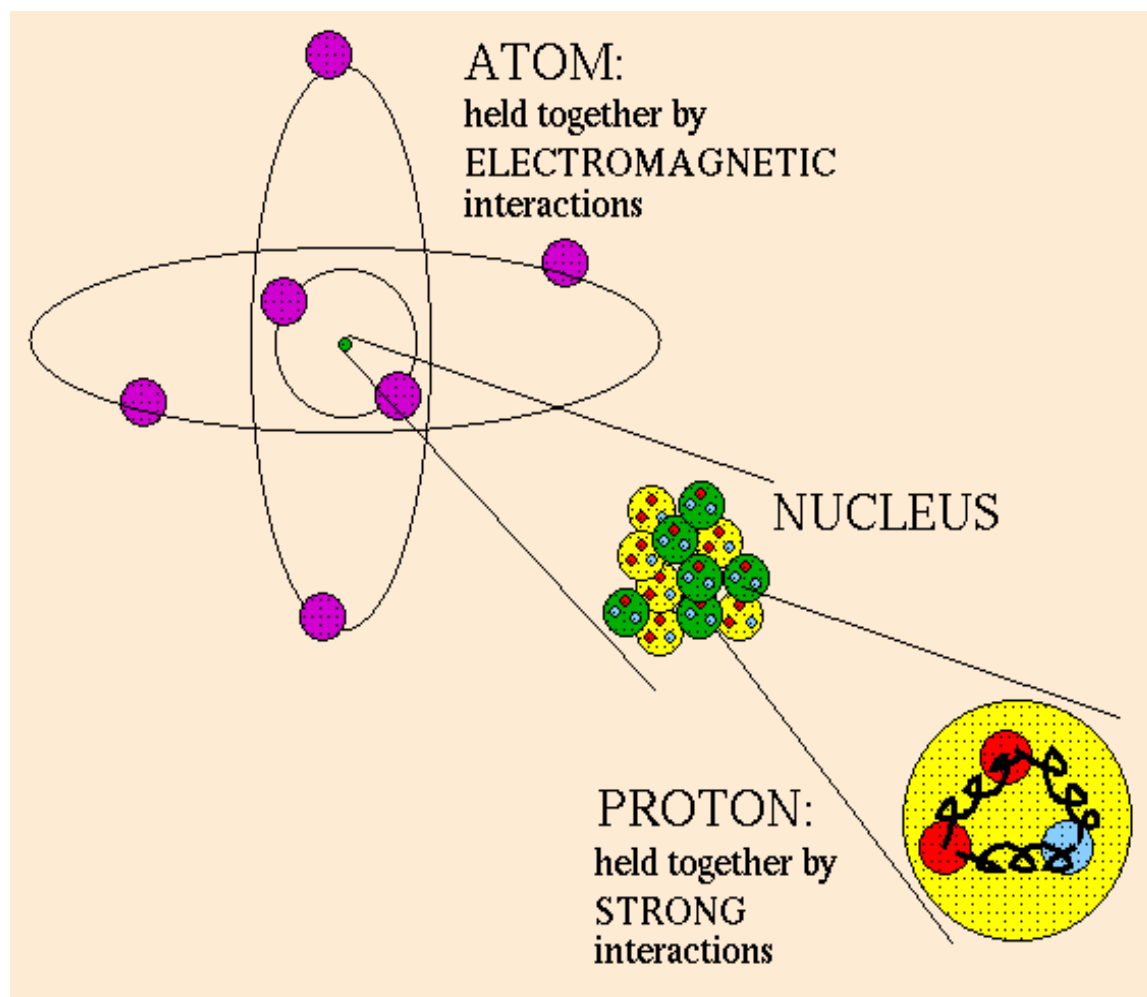
Lanthanide Series

Actinide Series

Phases
Solid
Liquid
Gas

(Mass Numbers in Parentheses are from the most stable of common isotopes.)

Ατομική και υπο-ατομική δόμηση της ύλης



Πώς τα ξέρουμε αυτά; Ακολουθεί λίγη ιστορία...

20th Century Elementary Particle Physics Timeline

Theoretical Breakthroughs

1900 Black body radiation spectrum (Planck)

1905 Light quanta (Einstein)
1905 Theory of special relativity (Einstein)

1913 Bohr's model of atom, momentum quantum (Bohr)

1925 Hypothesis of "no two fermions in identical states" (Pauli)

1925-26 Quantum Mechanics (Heisenberg, Schrodinger)

1928 Electron relativistic QM equation; antimatter (Dirac)

1930 Neutrino hypothesis (Pauli)

1934 Weak Force (Fermi)

1935 Nuclear Force and Yukawa particles (Yukawa)

Discoveries of Building Blocks

1897-1899 Discovery of electron (Thompson)

1911 Discovery of nucleus (Rutherford)

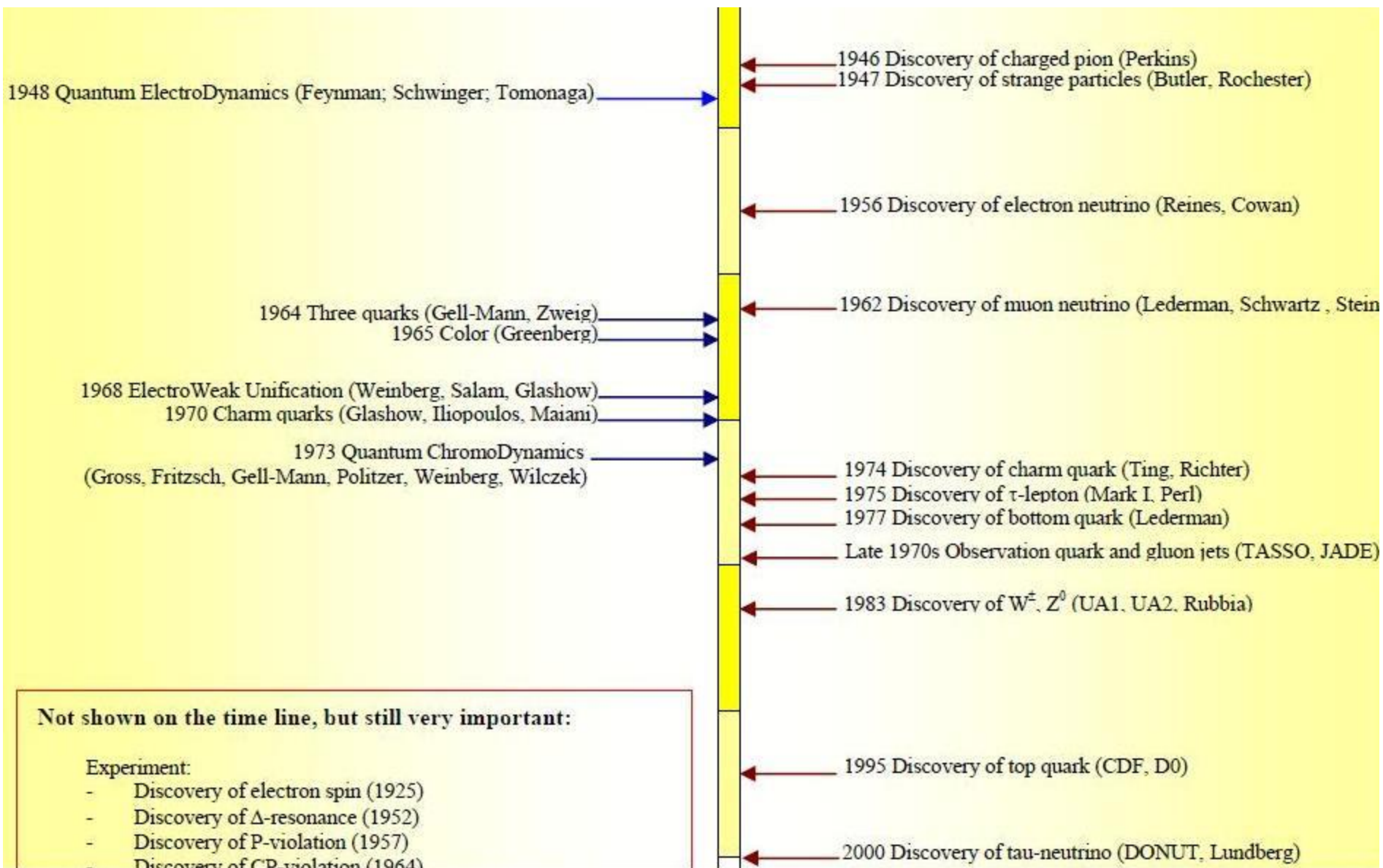
1919 Discovery of proton (Rutherford)

1923 "Discovery" of photon (Compton)

1932 Discovery of neutron (Chadwick)

1933 Discovery of antimatter, positron (Anderson)

1937 Discovery of muon (Anderson, Neddermeyer)



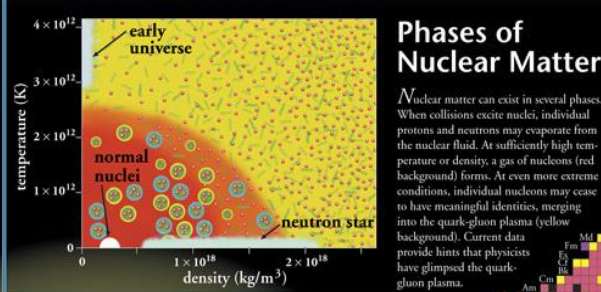
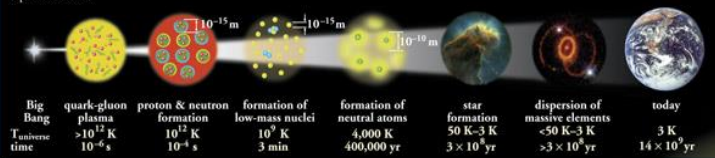
Nuclear Science

Nuclear Science is the study of the structure, properties, and interactions of the atomic nuclei. Nuclear scientists calculate and measure the masses, shapes, sizes, and decays of nuclei at rest and in collisions. They ask questions, such as: Why do nucleons stay in the nucleus? What combinations of protons and neutrons are possible? What happens when nuclei are compressed or rapidly rotated? What is the origin of the nuclei found on Earth?

Legend	electron (e^-)	quark	A_{mass} number 14
proton	positron (e^+)	gluon field	Z_{atomic} number 6
neutrino (ν)	antineutrino ($\bar{\nu}$)	photon (γ)	N_{neutron} number $A - Z$

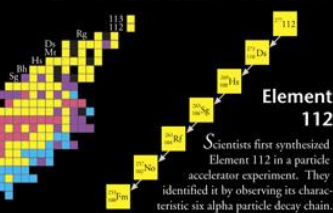
Expansion of the Universe

After the Big Bang, the universe expanded and cooled. At about 10^{-36} second, the universe consisted of a soup of quarks, gluons, electrons, and neutrinos. When the temperature of the Universe, T_{universe} , cooled to about 10^9 K, this soup coalesced into protons, neutrons, and electrons. As time progressed, some of the protons and neutrons formed deuterium, helium, and lithium nuclei. Still later, electrons combined with protons and these low-mass nuclei to form neutral atoms. Due to gravity, clouds of atoms contracted into stars, where hydrogen and helium fused into more massive chemical elements. Exploding stars (supernovae) form the most massive elements and disperse them into space. Our earth was formed from supernova debris.

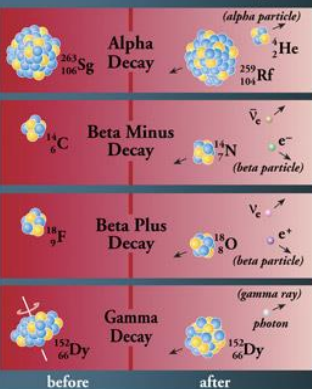


Unstable Nuclei

Stable nuclides form a narrow white band on the Chart of the Nuclides. Scientists produce unstable nuclides far from this band and study their decay, thereby learning about the extremes of nuclear conditions. In its present form, this chart contains about 2500 different nuclides. Nuclear theory predicts that there are at least 4000 more to be discovered with $Z \leq 113$.

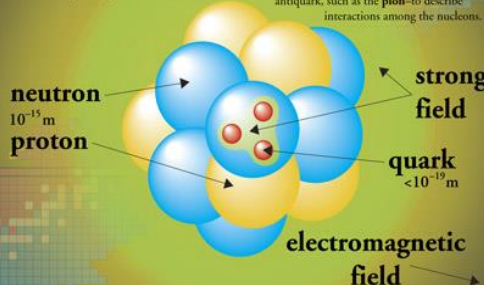


Radioactivity



Radioactive decay transforms a nucleus by emitting different particles. In alpha decay, the nucleus releases a ^4_2He nucleus—an alpha particle. In beta decay, the nucleus either emits an electron and an antineutrino (or a positron and neutrino) or captures an atomic electron and emits a neutrino. A positron is the name for the antiparticle of the electron. Antimatter is composed of anti-particles. Both alpha and beta decays change the original nucleus into a nucleus of a different chemical element. In gamma decay, the nucleus lowers its internal energy by emitting a photon—a gamma ray. This decay does not modify the chemical properties of the atom.

The Nucleus



In an atom, electrons range around the nucleus at distances typically up to 10,000 times the nuclear diameter. If the electron cloud were shown to scale, this chart would cover a small town.

Chart of the Nuclides

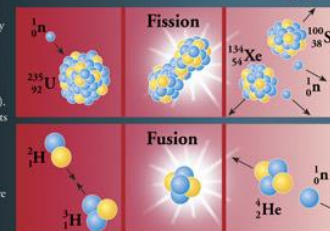
The Chart of the Nuclides presents in graphic form all known nuclei with atomic number, Z , and neutron number, N . Each nuclide is represented by a box colored according to its predominant decay mode. Magic numbers (N or $Z = 2, 8, 20, 28, 50, 82$ and 126) are indicated by a rectangle on the chart. They correspond to major closed shells and show regions of greater nuclear binding energy.



Color Key

- Stable
- Spontaneous fission
- Alpha particle emission
- Beta minus emission
- Beta plus emission or electron capture

Nuclear Energy



In the early stages of stellar evolution of our sun and other stars, hydrogen fuses to form helium, releasing energy in the form of photons (light) and neutrinos. During the later stages of stellar evolution, more massive nuclei up to and beyond uranium are synthesized by fusion. By measuring the number of neutrinos that come from the Sun, scientists recently have demonstrated that neutrinos must have a mass greater than zero.

Applications



Radioactive Dating

Naturally occurring radioactive isotopes such as ^{14}C are used to date objects that were once living, such as wood. For example, from a study of artifacts found at the site, scientists determined that Stonehenge was built nearly 4,000 years ago.



Smoke Detectors

Many smoke detectors use a small amount of the alpha emitter ^{241}Am to ionize the air. Smoke entering the detector reduces the current and sets off the alarm.



Nuclear Medicine

Radioactive isotopes, such as $^{99\text{m}}\text{Tc}$, ^{67}Ga , and ^{18}F , are commonly used in the diagnosis and treatment of disease. Positron emitters such as ^{18}F are used in Positron Emission Tomography (PET) to generate images of brain activity.



Nuclear Reactors

Nuclear reactors use the fusion of ^2_1H or ^3_1H nuclei to produce electric power. Reactors and most other nuclear applications generate radioactive waste; disposal of this waste is a subject of current research.

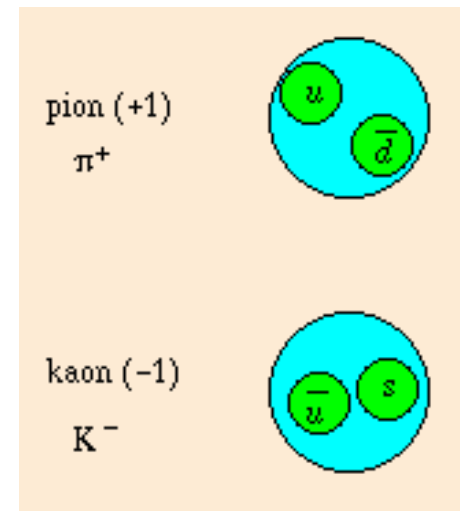
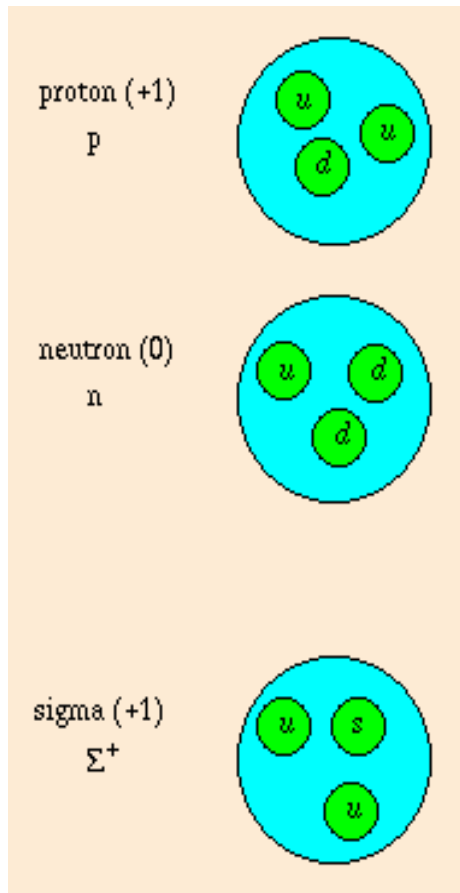


Magnetic Resonance Imaging

Magnetic Resonance Imaging (MRI) makes use of atomic transitions involving the magnetic field of a nucleus to study the local chemical environment. This technique accurately maps the density of hydrogen to produce three-dimensional images of the human body.

www.CPEPweb.org

Στοιχειώδη σωματίδια με εσωτερική δομή – αδρόνια (strong interactions)



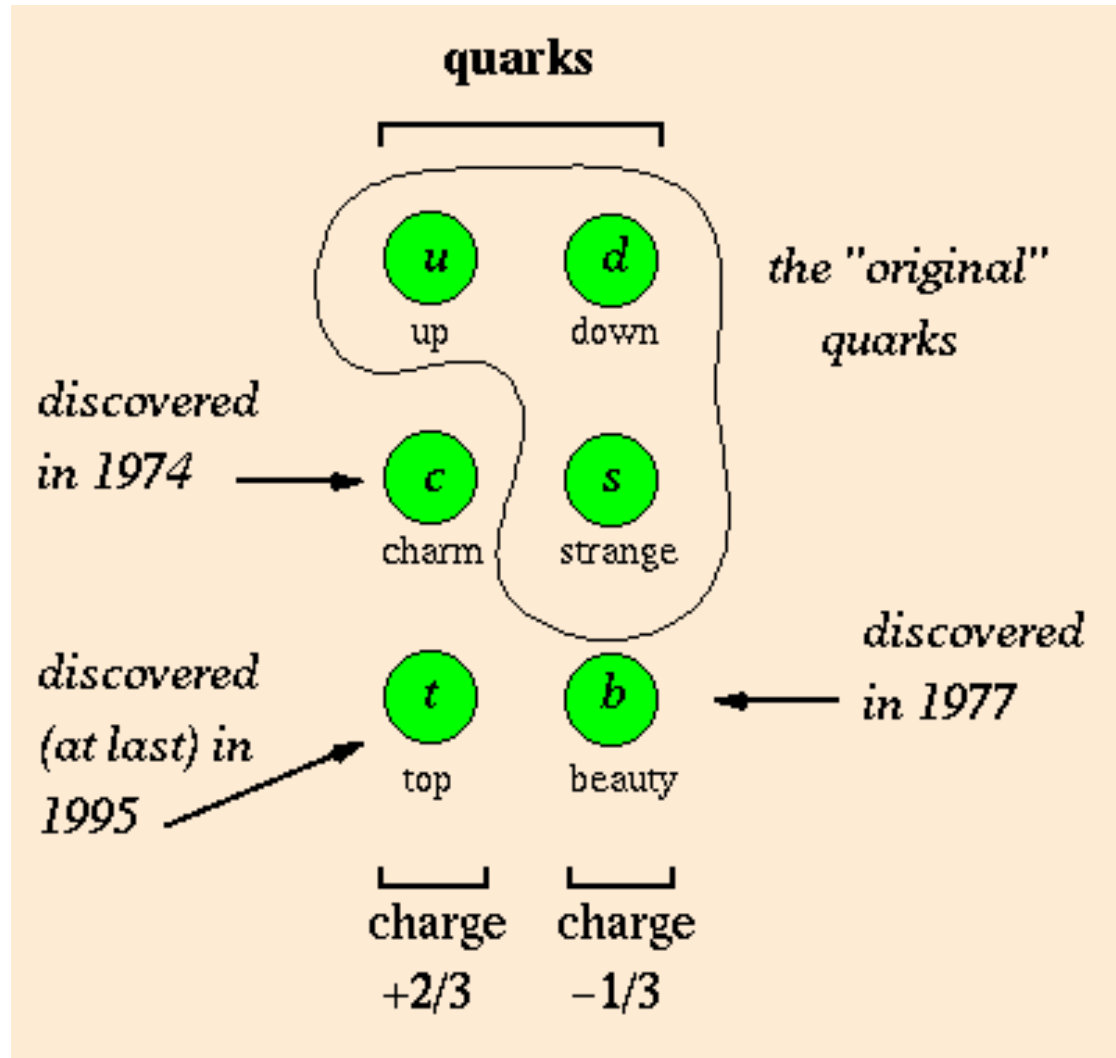
Μεσόνια:

αδρόνια με 1 quark και 1 antiquark

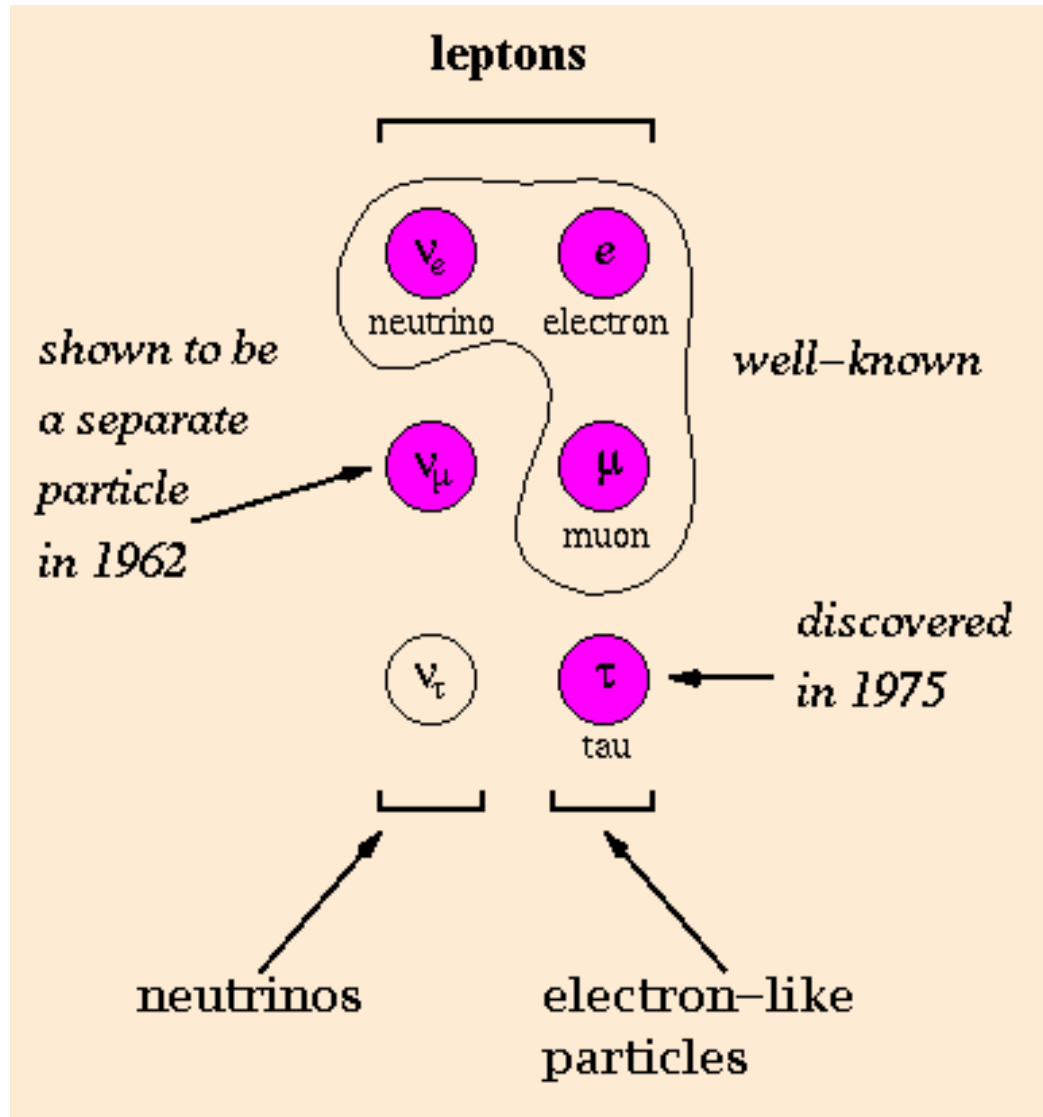
Βαρυόνια:

αδρόνια με 3 quarks ή 3 antiquark

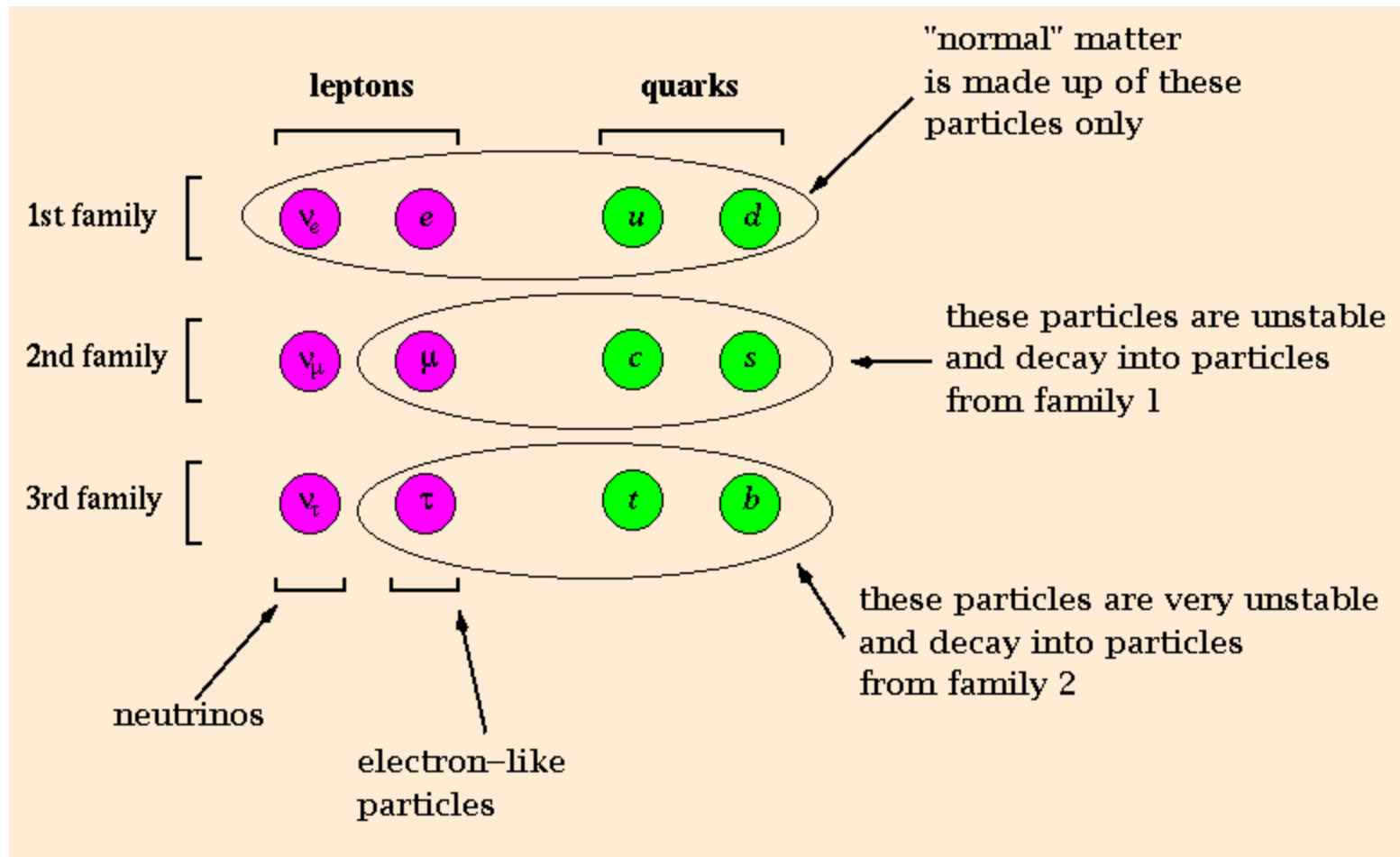
The 6 quarks



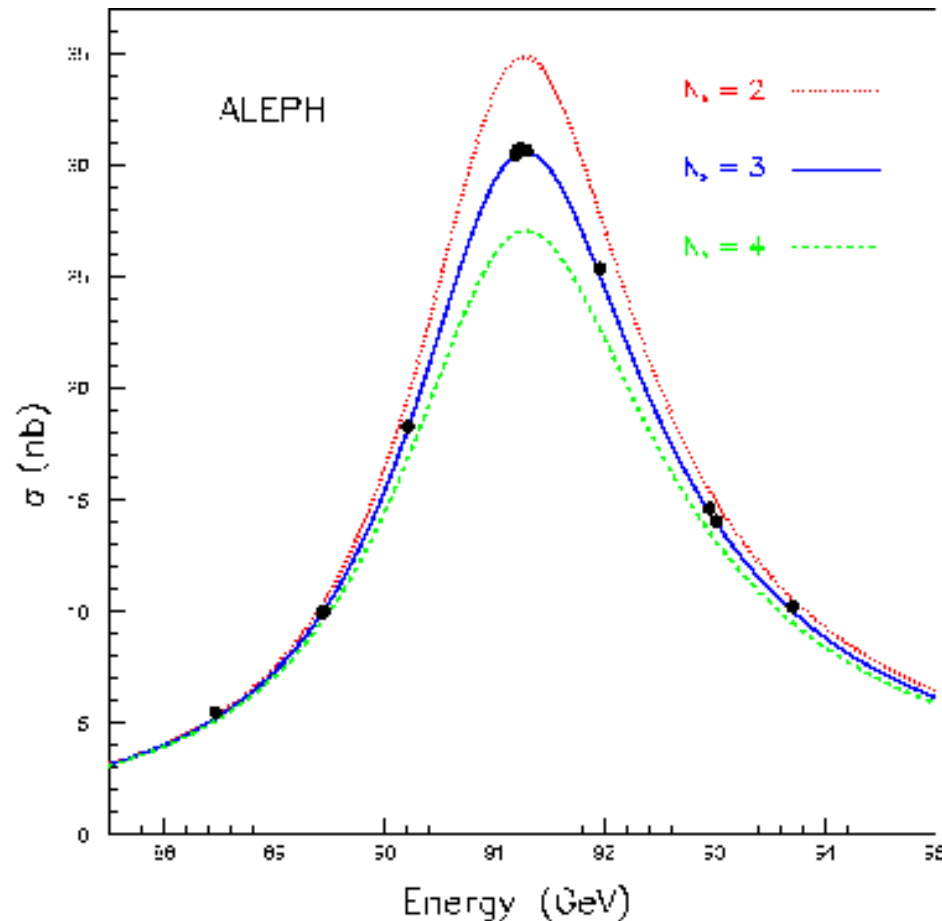
The 6 leptons



Λεπτόνια και quark σε 3 οικογένειες

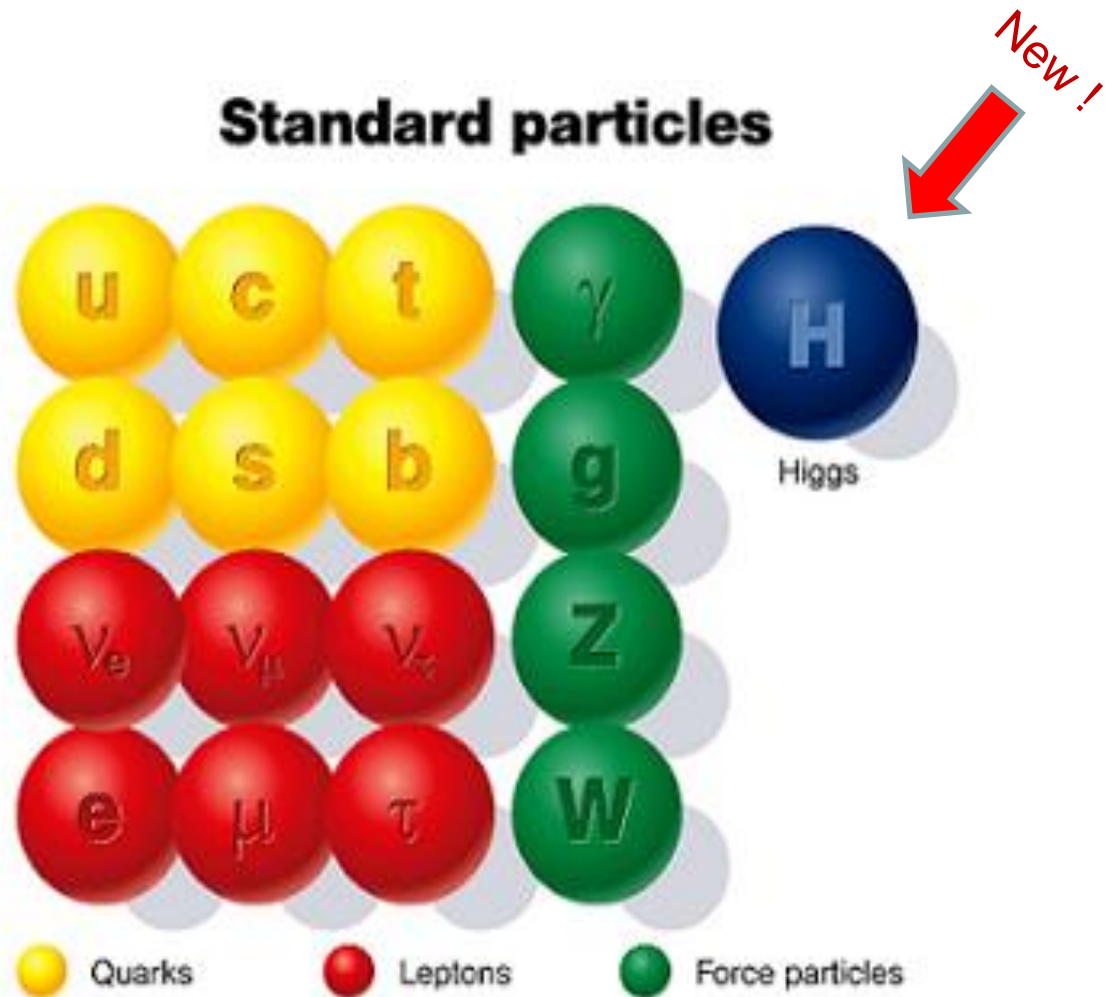


Measurement of Z hadronic decays, versus energy,
with predicted rates if there were 1, 2 or 3 types of neutrino
(ALEPH collaboration)



Τα δομικά στοιχεία της ύλης

(σε υποατομικό - υποπυρηνικό επίπεδο)



Standard Model of FUNDAMENTAL PARTICLES AND INTERACTIONS

The Standard Model summarizes the current knowledge in Particle Physics. It is the quantum theory that includes the theory of strong interactions (quantum chromodynamics or QCD) and the unified theory of weak and electromagnetic interactions (electroweak). Gravity is included on this chart because it is one of the fundamental interactions even though not part of the "Standard Model."

FERMIONS

matter constituents
spin = 1/2, 3/2, 5/2, ...

Leptons spin = 1/2			Quarks spin = 1/2		
Flavor	Mass GeV/c ²	Electric charge	Flavor	Approx. Mass GeV/c ²	Electric charge
$\bar{\nu}_e$ electron neutrino	$<1 \times 10^{-8}$	0	u up	0.003	2/3
e electron	0.000511	-1	d down	0.006	-1/3
$\bar{\nu}_\mu$ muon neutrino	<0.0002	0	c charm	1.3	2/3
μ muon	0.106	-1	s strange	0.1	-1/3
$\bar{\nu}_\tau$ tau neutrino	<0.02	0	t top	175	2/3
τ tau	1.7771	-1	b bottom	4.3	-1/3

Spin is the intrinsic angular momentum of particles. Spin is given in units of \hbar , which is the quantum unit of angular momentum, where $\hbar = h/2\pi = 6.58 \times 10^{-25}$ GeV s = 1.05×10^{-34} J s.

Electric charges are given in units of the proton's charge. In SI units the electric charge of the proton is 1.60×10^{-19} coulombs.

The **energy** unit of particle physics is the electronvolt (eV), the energy gained by one electron in crossing a potential difference of one volt. **Masses** are given in GeV/c² (remember $E = mc^2$), where $1 \text{ GeV} = 10^9 \text{ eV} = 1.60 \times 10^{-10}$ joule. The mass of the proton is $0.938 \text{ GeV}/c^2 = 1.67 \times 10^{-27} \text{ kg}$.

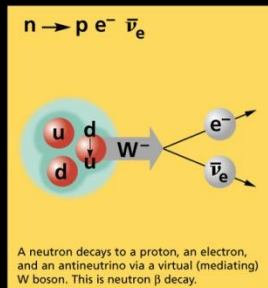
Baryons qqq and Antibaryons $\bar{q}\bar{q}\bar{q}$					
Baryons are fermionic hadrons. There are about 120 types of baryons.					
Symbol	Name	Quark content	Electric charge	Mass GeV/c ²	Spin
p	proton	uud	1	0.938	1/2
\bar{p}	anti-proton	$\bar{u}\bar{u}\bar{d}$	-1	0.938	1/2
n	neutron	udd	0	0.940	1/2
Λ	lambda	uds	0	1.116	1/2
Ω^-	omega	sss	-1	1.672	3/2

Matter and Antimatter

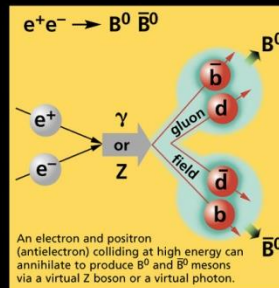
For every particle type there is a corresponding antiparticle type, denoted by a bar over the particle symbol (unless + or - charge is shown). Particle and antiparticle have identical mass and spin but opposite charges. Some electrically neutral bosons (e.g., Z^0 , γ , and $\eta_c = c\bar{c}$, but not $K^0 = d\bar{s}$) are their own antiparticles.

Figures

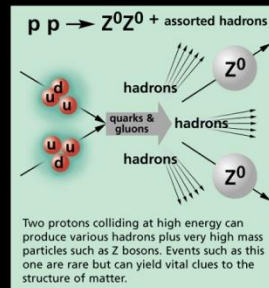
These diagrams are an artist's conception of physical processes. They are not exact and have no meaningful scale. Green shaded areas represent the cloud of gluons or the gluon field, and red lines the quark paths.



A neutron decays to a proton, an electron, and an antineutrino via a virtual (mediating) W boson. This is neutron β decay.

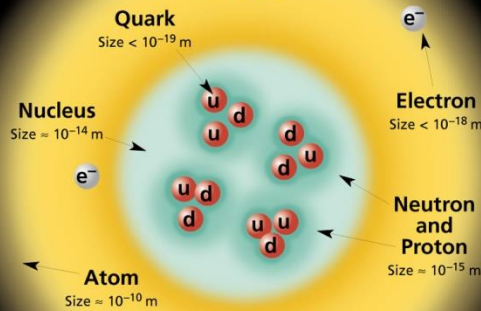


An electron and positron (antielectron) colliding at high energy can annihilate to produce B^0 and \bar{B}^0 mesons via a virtual Z boson or a virtual photon.



Two protons colliding at high energy can produce various hadrons plus very high mass particles such as Z bosons. Events such as this one are rare but can yield vital clues to the structure of matter.

Structure within the Atom



If the protons and neutrons in this picture were 10 cm across, then the quarks and electrons would be less than 0.1 mm in size and the entire atom would be about 10 km across.

BOSONS

force carriers
spin = 0, 1, 2, ...

Unified Electroweak spin = 1			Strong (color) spin = 1		
Name	Mass GeV/c ²	Electric charge	Name	Mass GeV/c ²	Electric charge
γ photon	0	0	g gluon	0	0
W^-	80.4	-1			
W^+	80.4	+1			
Z^0	91.187	0			

Color Charge

Each quark carries one of three types of "strong charge," also called "color charge." These charges have nothing to do with the colors of visible light. There are eight possible types of color charge for gluons. Just as electrically-charged particles interact by exchanging photons, in strong interactions color-charged particles interact by exchanging gluons. Leptons, photons, and W and Z bosons have no strong interactions and hence no color charge.

Quarks Confined in Mesons and Baryons

One cannot isolate quarks and gluons; they are confined in color-neutral particles called **hadrons**. This confinement (binding) results from multiple exchanges of gluons among the color-charged constituents. As color-charged particles (quarks and gluons) move apart, the energy in the color-force field between them increases. This energy eventually is converted into additional quark-antiquark pairs (see figure below). The quarks and antiquarks then combine into hadrons; these are the particles seen to emerge. Two types of hadrons have been observed in nature: **mesons** $q\bar{q}$ and **baryons** qqq .

Residual Strong Interaction

The strong binding of color-neutral protons and neutrons to form nuclei is due to residual strong interactions between their color-charged constituents. It is similar to the residual electrical interaction that binds electrically neutral atoms to form molecules. It can also be viewed as the exchange of mesons between the hadrons.

PROPERTIES OF THE INTERACTIONS

Property \ Interaction	Gravitational	Weak	Electromagnetic	Strong	
		(Electroweak)		Fundamental	Residual
Acts on:	Mass - Energy	Flavor	Electric Charge	Color Charge	See Residual Strong Interaction Note
Particles experiencing:	All	Quarks, Leptons	Electrically charged	Quarks, Gluons	Hadrons
Particles mediating:	Graviton (not yet observed)	$W^+ W^- Z^0$	γ	Gluons	Mesons
Strength relative to electromag for two u quarks at:		0.8	1	25	Not applicable to quarks
for two u quarks at: 10^{-18} m	10^{-41}	10^{-4}	1	60	
for two protons in nucleus $3 \times 10^{-17} \text{ m}$	10^{-36}	10^{-7}	1	Not applicable to hadrons	20

Mesons $q\bar{q}$					
Mesons are bosonic hadrons. There are about 140 types of mesons.					
Symbol	Name	Quark content	Electric charge	Mass GeV/c ²	Spin
π^+	pion	u\bar{d}	+1	0.140	0
K^-	kaon	s\bar{u}	-1	0.494	0
ρ^+	rho	u\bar{d}	+1	0.770	1
B^0	B-zero	d\bar{b}	0	5.279	0
η_c	eta-c	c\bar{c}	0	2.980	0

The Particle Adventure

Visit the award-winning web feature *The Particle Adventure* at <http://ParticleAdventure.org>

This chart has been made possible by the generous support of:

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U.S. National Science Foundation
Lawrence Berkeley National Laboratory
Stanford Linear Accelerator Center
American Physical Society, Division of Particles and Fields
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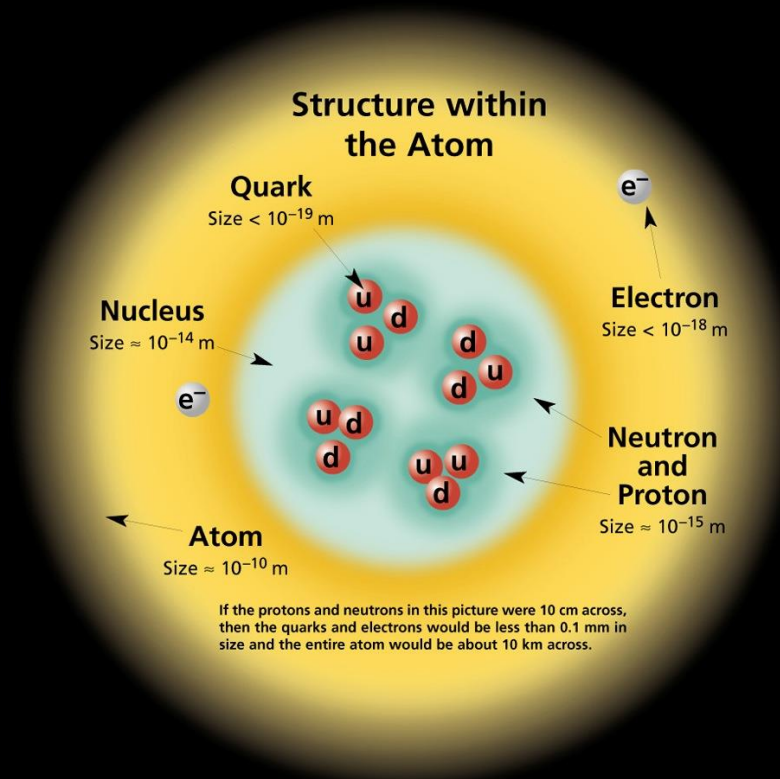
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The Standard Model of Particles and Fields

Main postulates of SM:

- 1) Basic constituents of matter are *quarks* and *leptons* (fermions i.e. spin 1/2).
- 2) They interact by means of gauge bosons (spin 1).
- 3) Quarks and leptons are subdivided into 3 *generations*.



FERMIONS

matter constituents
spin = 1/2, 3/2, 5/2, ...

Leptons spin = 1/2		
Flavor	Mass GeV/c ²	Electric charge
ν_e electron neutrino	$<1 \times 10^{-8}$	0
e electron	0.000511	-1
ν_μ muon neutrino	<0.0002	0
μ muon	0.106	-1
ν_τ tau neutrino	<0.02	0
τ tau	1.7771	-1

Quarks spin = 1/2		
Flavor	Approx. Mass GeV/c ²	Electric charge
u up	0.003	2/3
d down	0.006	-1/3
c charm	1.3	2/3
s strange	0.1	-1/3
t top	175	2/3
b bottom	4.3	-1/3

Standard Model
does not
explain neither
appearance of
the mass
nor the reason
for existence of
3 generations

Spin is the intrinsic angular momentum of particles. Spin is given in units of \hbar , which is the quantum unit of angular momentum, where $\hbar = h/2\pi = 6.58 \times 10^{-25} \text{ GeV s} = 1.05 \times 10^{-34} \text{ J s}$.

Electric charges are given in units of the proton's charge. In SI units the electric charge of the proton is $1.60 \times 10^{-19} \text{ coulombs}$.

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Forces of Nature

Forces are being carried by specific particles, called *gauge* [*gejdz*] *bosons*.

Property \ Interaction	Gravitational	Weak (Electroweak)	Electromagnetic	Strong	
				Fundamental	Residual
Acts on:	Mass – Energy	Flavor	Electric Charge	Color Charge	See Residual Strong Interaction Note
Particles experiencing:	All	Quarks, Leptons	Electrically charged	Quarks, Gluons	Hadrons
Particles mediating:	Graviton (not yet observed)	W^+ W^- Z^0	γ	Gluons	Mesons
Strength relative to electromag for two u quarks at: $\left\{ \begin{array}{l} 10^{-18} \text{ m} \\ 3 \times 10^{-17} \text{ m} \end{array} \right.$ for two protons in nucleus	10^{-41}	0.8	1	25	Not applicable to quarks
	10^{-41}	10^{-4}	1	60	
	10^{-36}	10^{-7}	1	Not applicable to hadrons	20

⇒ Electromagnetic and weak forces can be described by a single theory ⇒ the “*Electroweak Theory*” was developed in 1960s (Glashow, Weinberg, Salam).

⇒ Theory of strong interactions appeared in 1970s: “*Quantum Chromodynamics*” (QCD).

⇒ The “*Standard Model*” (SM) combines both.

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spin = 0, 1, 2, ...

Unified Electroweak spin = 1		
Name	Mass GeV/c ²	Electric charge
γ photon	0	0
W^-	80.4	-1
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Z^0	91.187	0

Strong (color) spin = 1		
Name	Mass GeV/c ²	Electric charge
g gluon	0	0

Color Charge

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One cannot isolate quarks and gluons; they are confined in color-neutral particles called **hadrons**. This confinement (binding) results from multiple exchanges of gluons among the color-charged constituents. As color-charged particles (quarks and gluons) move apart, the energy in the color-force field between them increases. This energy eventually is converted into additional quark-antiquark pairs (see figure below). The quarks and antiquarks then combine into hadrons; these are the particles seen to emerge. Two types of hadrons have been observed in nature: **mesons** $q\bar{q}$ and **baryons** qqq .

Residual Strong Interaction

The strong binding of color-neutral protons and neutrons to form nuclei is due to residual strong interactions between their color-charged constituents. It is similar to the residual electrical interaction that binds electrically neutral atoms to form molecules. It can also be viewed as the exchange of mesons between the hadrons.

Matter and Antimatter

For every particle type there is a corresponding antiparticle type, denoted by a bar over the particle symbol (unless + or – charge is shown). Particle and antiparticle have identical mass and spin but opposite charges. Some electrically neutral bosons (e.g., Z^0 , γ , and $\eta_c = c\bar{c}$, but not $K^0 = d\bar{s}$) are their own antiparticles.

Baryons qqq and Antibaryons $\bar{q}\bar{q}\bar{q}$

Baryons are fermionic hadrons.
There are about 120 types of baryons.

Symbol	Name	Quark content	Electric charge	Mass GeV/c^2	Spin
p	proton	uud	1	0.938	1/2
\bar{p}	anti-proton	$\bar{u}\bar{u}\bar{d}$	–1	0.938	1/2
n	neutron	udd	0	0.940	1/2
Λ	lambda	uds	0	1.116	1/2
Ω^-	omega	sss	–1	1.672	3/2

Mesons $q\bar{q}$

Mesons are bosonic hadrons.
There are about 140 types of mesons.

Symbol	Name	Quark content	Electric charge	Mass GeV/c^2	Spin
π^+	pion	$u\bar{d}$	+1	0.140	0
K^-	kaon	$s\bar{u}$	-1	0.494	0
ρ^+	rho	$u\bar{d}$	+1	0.770	1
B^0	B-zero	$d\bar{b}$	0	5.279	0
η_c	eta-c	$c\bar{c}$	0	2.980	0